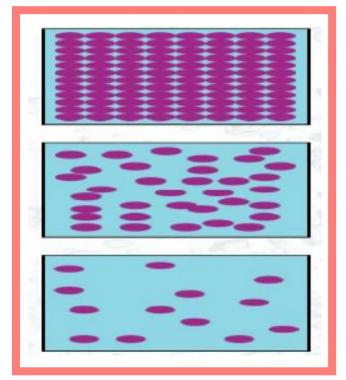


Key Vocabulary and Spellings		
1	solid	An object firm in shape
2	liquid	A substance that moves freely
3	gas	A substance that moves freely and may not be seen
4	state	A condition of something
5	evaporate	Liquid changing to gas
6	condensation	Gas changing to liquid
7	temperature	How hot or cold something is
8	degrees	A measure of temperature
9	particle	A minute part of matter
10	properties	How something can be classified by

Top 5 Topic Facts

- Particles in solids are closely packedmeaning many solids are not malleable.
- Particles in liquids are more spreadout, meaning liquids can move and change shape.
- Particles in gases are spread out
 even further, meaning they move freely and cannot always be seen.
- 4 Liquids boils at 100 degrees Celsius, liquids freeze at 0 degrees Celsius.
- Evaporation is water changing stateto a gas. Condensation is when gas changes into a liquid.



	Key Investigations/Questions		
1	1	What are solids, liquids and gases and how can we identify them?	
2	2	What conditions are required to change the state of an object or substance?	
	3	What is condensation and evaporation and how can this be tested?	